

BME303

Third Semester B.E./B.Tech Degree Examination, Dec.2023/Jan.2024 **Material Science and Engineering**

Time: 3 hrs.

Max. Marks: 100

Note: 1. Answer any FIVE full questions, choosing ONE full question from each module. 2. M: Marks, L: Bloom's level, C: Course outcomes.

	Module -1	M	L	C
a.	Calculate the APF for FCC and BCC unit cell in crystal structure.	10	L3	CO ₁
2.1	Enumerate the type of crystal imperfections and explain briefly with a	10	L2	CO ₁
	suitable sketch grain boundary and twin boundry defects.			
8	OR			~~1
a.	Explain briefly with suitable sketches the plastic deformation by Slip and	10	L2	CO ₁
	Twinning.	_	T O	001
b.		5	L2	CO ₁
100	ii) Coordination number.	5	1.2	CO1
c.	Molybdenum has BCC and a density of 10.2 × 10° kg/m ³ . Calculate its	3	LS	COI
	atomic radius. The atomic weight of molybdenum is 95.94gm/mol.	200		
200	$N_A = 6.023 \times 10^{23}$ atoms/mol.	10 0	2 3 7 2 1	
			- 1000	
	Module – 2	10	12	CO2
a.		10	LL	002
14	explain briefly.	10	12	CO2
b.	State and explain briefly the Fick's 1 and 2 law of diffusion.	LU		COL
		08	1.2	CO2
a.	Explain briefly with a neat sketch the eutectic system of two components	00		
1	Completely soluble in liquid state and partially soluble in solid state.	12	L2	CO2
b.	or the diagram also show the three invariant reactions	1	7.3	
3.00	on the diagram also show the three mountain reactions.			2
	Module – 3			
0	Explain briefly mechanism of solidification with suitable sketches.	10	L2	CO.
-	With a suitable sketch explain normalizing heat treatment process.	10	L2	CO.
D.		15.	1 4	1.47
a.	Draw a neat labeled Time-Temperature Transformation [TTT] diagram for	10	13	CO.
a.	Blaw a near lasered 1 mile 1 mile		Lo	10.00
9.	Futectoid steel (0.8%C) and explain briefly.		LJ	
b.	Eutectoid steel (0.8%C) and explain briefly. With a neat sketch briefly explain Aaustemperign and Martempering heat	10	L2	
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	With a neat sketch briefly explain Aaustemperign and Martempering heat treatment process. Module – 4 With a flow chart, explain briefly the powder metallurgy process and its applications.	10	L2	CO
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a.	With a neat sketch briefly explain Aaustemperign and Martempering heat treatment process. Module – 4 With a flow chart, explain briefly the powder metallurgy process and its applications. Enumerate the different powder production methods, with suitable sketch explain atomization method.	10	L2	CO
a.	With a neat sketch briefly explain Aaustemperign and Martempering heat treatment process. Module – 4 With a flow chart, explain briefly the powder metallurgy process and its applications. Enumerate the different powder production methods, with suitable sketch explain atomization method. OR	10 10 10	L2 L2	CO.
a.	With a neat sketch briefly explain Aaustemperign and Martempering heat treatment process. Module – 4 With a flow chart, explain briefly the powder metallurgy process and its applications. Enumerate the different powder production methods, with suitable sketch explain atomization method. OR Explain briefly thermal spray coating with suitable sketch. Mention the	10	L2 L2	CO.
a. b.	With a neat sketch briefly explain Aaustemperign and Martempering heat treatment process. Module – 4 With a flow chart, explain briefly the powder metallurgy process and its applications. Enumerate the different powder production methods, with suitable sketch explain atomization method. OR	10 10 10	L2	CO4
	a. b. a. b. a. b.	 b. Enumerate the type of crystal imperfections and explain briefly with a suitable sketch grain boundary and twin boundry defects. OR a. Explain briefly with suitable sketches the plastic deformation by Slip and Twinning. b. Define the following terms: i) Unit cell and space lattice ii) Coordination number. c. Molybdenum has BCC and a density of 10.2 × 10³ kg/m³. Calculate its atomic radius. The atomic weight of molybdenum is 95.94gm/mol. N_A = 6.023 × 10²³ atoms/mol. Module -2 a. Draw neatly the solid solution binary phase diagram of a Ni-Cu system and explain briefly. b. State and explain briefly the Fick's 1st and 2nd law of diffusion. OR a. Explain briefly with a neat sketch the eutectic system of two components completely soluble in liquid state and partially soluble in solid state. b. Draw a neat sketch of iron-carbon equilibrium diagram and show all phases on the diagram also show the three invariant reactions. Module -3 a. Explain briefly mechanism of solidification with suitable sketches. b. With a suitable sketch explain normalizing heat treatment process. 	b. Enumerate the type of crystal imperfections and explain briefly with a suitable sketch grain boundary and twin boundry defects. OR a. Explain briefly with suitable sketches the plastic deformation by Slip and Twinning. b. Define the following terms: i) Unit cell and space lattice ii) Coordination number. c. Molybdenum has BCC and a density of 10.2 × 10 ³ kg/m ³ . Calculate its atomic radius. The atomic weight of molybdenum is 95.94gm/mol. N _A = 6.023 × 10 ²³ atoms/mol. Module -2 a. Draw neatly the solid solution binary phase diagram of a Ni-Cu system and explain briefly. b. State and explain briefly the Fick's 1 st and 2 nd law of diffusion. OR a. Explain briefly with a neat sketch the cutectic system of two components completely soluble in liquid state and partially soluble in solid state. b. Draw a neat sketch of iron-carbon equilibrium diagram and show all phases on the diagram also show the three invariant reactions. Module - 3 a. Explain briefly mechanism of solidification with suitable sketches. 10 NoR	b. Enumerate the type of crystal imperfections and explain briefly with a suitable sketch grain boundary and twin boundry defects. OR a. Explain briefly with suitable sketches the plastic deformation by Slip and Twinning. b. Define the following terms: i) Unit cell and space lattice ii) Coordination number. c. Molybdenum has BCC and a density of 10.2 × 10³ kg/m³. Calculate its atomic radius. The atomic weight of molybdenum is 95.94gm/mol. N _A = 6.023 × 10²³ atoms/mol. Module -2 a. Draw neatly the solid solution binary phase diagram of a Ni-Cu system and explain briefly. b. State and explain briefly the Fick's 1st and 2nd law of diffusion. OR a. Explain briefly with a neat sketch the cutectic system of two components completely soluble in liquid state and partially soluble in solid state. b. Draw a neat sketch of iron-carbon equilibrium diagram and show all phases on the diagram also show the three invariant reactions. Module - 3 a. Explain briefly mechanism of solidification with suitable sketches. 10 L2 OR A. Explain briefly mechanism of solidification with suitable sketches. 10 L2 OR