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I Semester M.Sc. Degree Examination, June/July - 2022

CHEMISTRY

Organic Chemistry-I

(CBCS Syllabus Scheme - 2019-20 Onwards)

Paper: CH-102

Time: 3 Hours

Instructions to Candidates:

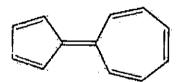
- Answer question No. 1 any Ten and Tive from remaining questions.
- 2) Figures to the right indicate marks.

Answer any TEN of the following:

 $(10 \times 2 = 20)$

MSximum Marks: 70

- 1. a) What is cross-conjugation? Give an example.
 - b) Write two resonance forms for the following compound. Comment on their stability and aromaticity.



- c) What are crown-ethers? Give any two examples.
- d) Give reasons why triplet methylene is more stable than singlet?
- e) Outline any two methods of generation of carbanions.
- f) Arrange the following separately as hard and soft acids and bases.

$$H_2O, Na^+, OH^-, Cu^+, SCN^-, H^-$$

g) Assign R/S configuration to the following chiral compounds.

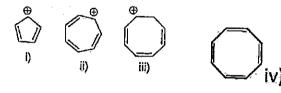
$$CH_3$$
 CH_3 H OH H OH CH_3 CH_3 CH_3

- h) Outline prochirality with suitable example.
- i) Comment on the optical activity of allenes.
- j) Sketch the structures of gentiobiose and meliobiose.
- k) Give the IUPAC names and numbering for the following compounds.





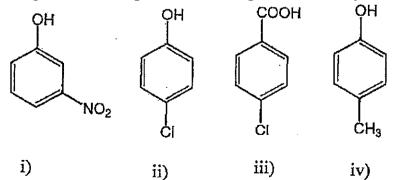
- 1) What are deoxysugars? Give an example.
- **2.** a) Comment on the aromaticity of the following:



- b) Account the aromaticity of the alternant and non-alternant hydrocarbons, with suitable examples. (4+6=10)
- 3. a) Sketch and explain the potential energy diagrams, transition state and intermediates in an S_{N^1} reaction.
 - b) In spite of having $[4n+2\pi]$ electrons [10] annulene is not aromatic? Why?
 - c) Citing examples, highlight optical activity of allenes.

(4+3+3=10)

- 4. a) Discuss briefly the generation, stability and chemical reaction of carbenes.
 - b) How the reaction mechanism is determined by isotopic labelling? Explain with suitable example. (6+4=10)
- 5. a) Arrange the following in the increasing order of acidity. Justify.



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- b) Write short notes on:
 -) Taft equation.
 - ii) SEZ reaction.
 - iii) Anchimeric effect.

(4+6=10)

6. a) Predict the stereochemistry of the product using Cram's rule with suitable explanation.

- b) Write a note on conformational analysis of 1-methyl cyclohexane. (6+4=10)
- 7. a) Write the Fischer, Sawhorse, Newman and Flying wedge representation structures for 2-bromo-3-butanol.
 - o) Outline the structural elucidation of sucrose.
- 8. a) Describe briefly the photosynthesis of carbohydrates.
 - b) Discuss any one synthesis and any one reaction of the following heterocycles:
 - i) Isoxazole.
 - ii) Benzimidazole.

(4+6=10)

(5+5=10)