Rajiv Gandhi University of Health Sciences, Karnataka

I Year Pharm-D Degree Examination - NOVEMBER 2015

Time: Three Hours Max. Marks: 70 Marks

PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. CODE: 2877

Your answers should be specific to the questions asked Draw neat, labeled diagrams wherever necessary

LONG ESSAYS (Answer any two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. Define and give the mechanism of: (a) Cannizzaro reaction (b) Aldol condensation.
- 2. State the Anti-Markovnikov's rule. Write the mechanism of addition of HBr to propene in the absence and presence of peroxide.
- 3. Write the mechanism of Friedel Craft's alkylation and acylation with their limitations.

SHORT ESSAYS (Answer any six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. Write the structure and medicinal uses of Benzyl benzoate, Saccharin sodium.
- 5. Define and classify isomerism with suitable examples.
- 6. Discuss about SN₁ mechanism and stereochemistry.
- 7. Write the effect of temperature in 1,2-addition and 1,4 addition.
- 8. "Write the synthesis of Dimercaprol and mention the use".
- 9. Explain the mechanism of dehydration of Alcohols.
- 10. What is Bayer's strain theory and explain with examples.
- 11. Give the mechanism of Benzoin condensation.

SHORT ANSWERS $10 \times 2 = 20 \text{ Marks}$

- 12. Outline the conversion of carboxylic acid to ester.
- 13. Define keto-enol Isomerism with examples.
- 14. Define Nucleophiles with examples.
- 15. Name the following compounds:

$$\begin{array}{c} \text{CH}_3 \\ \text{I} \\ \text{(a) CH}_3\text{-CH}_2\text{-CH}_2\text{-COOH} \\ \text{(b) CH}_3\text{-C-CH}_3 \\ \text{I} \\ \text{CH}_3 \end{array}$$

- 16. State the Lowry Bronsted theory of Acids and Bases.
- 17. Which is more stable trans-2-butane or cis-2-butane? Why?
- 18. What is Homolysis and Heterolysis?
- 19. Comment phenol is Acidic but not ethanol.
- 20. Define hyper conjugation.
- 21. Outline the conversion of Aniline to Benzoic Acid.
