## Rajiv Gandhi University of Health Sciences, Karnataka First Semester B.Pharm Degree Examination – JUNE-2019

Time: Three Hours Max. Marks: 75 Marks

## Pharmaceutical Inorganic Chemistry Q.P. CODE: 5004

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

## LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$ 

- 1. With the help of a neat labelled diagram explain the principle procedure involved in the Arsenic limit test.
- 2. What are anti microbials? Explain their mechanism. Write the preparation and uses of A) Chlorinated lime B) Hydrogen peroxide.
- 3. Write the principle reaction and procedure involved in the assay of copper sulphate and Ammonium chloride.

## **SHORT ESSAYS (Answer any Seven)**

 $7 \times 5 = 35 \text{ Marks}$ 

- 4. Define astringent. Write the preparation of Zinc sulphate.
- 5. Define buffers. Derive Henderson-Hasselbatch equation for buffers.
- 6. Write a not eon physiological acid base balance.
- 7. Write the principle and the reaction for iron limit test. Give the role of ammonia and citric acid.
- 8. What is saline cathartic? Write its mechanism of action. Give the preparation of any one cathartic.
- 9. Write a note on radio pharmaceutics.
- 10. Write a note on antidotes.
- 11. Define dentifrices. Discuss the role of fluoride in dentistry
- 12. Write a note on storage and handling of a radiopharmaceuticals.

SHORT ANSWERS  $10 \times 2 = 20 \text{ Marks}$ 

- 13. Give the method of preparation of kaolin.
- 14. Write the reaction for lead limit test.
- 15. Composition of ORS.
- 16. Write the preparation of calcium carbonate.
- 17. Combination of antacids.
- 18. Define haematinics with example.
- 19. Write the use of Thioglycolic acid in iron limit test.
- 20. Define Expectorant with example.
- 21. Write the limit test for sulphate.
- 22. Define emetics with example.

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